

SYNTHETIC STUDY AND BIOLOGICAL EVALUATION OF 1-TETRA-*O*-BENZOYL- β -D-GLUCOPYRANOSYL-3-SUBSTITUTED THIOCARBAMIDES

Kishor N. Puri *and Gajanan V. Korpe

P. G. Department of Chemistry, Shri Shivaji College, Akola-444 001(M.S.), India.

*E-mail: knpuri2008@rediffmail.com

ABSTRACT

Carbohydrate play vital role in vast array of biological processes particularly there are many advantages for example carbohydrate based drug show low toxicity and immunogenicity thus it is interesting to synthesize the carbohydrate base derivatives. In the present investigation we here in report series of 1-tetra-*O*-benzoyl- β -D-glucopyranosyl-3-substituted thiocarbamides which has been synthesized by the interaction of tetra-*O*-benzoyl- β -D-glucopyranosyl isothiocyanate and various alkyl amines. The structures of the newly synthesized compounds have been established on the basis of usual chemical transformations and IR, ¹HNMR and Mass spectral studies. All the synthesized compounds have been evaluated for their in vitro anti-bacterial and anti-fungal activity against different bacteria and fungi by agar diffusion method.

Keywords: *N*-glucosides, alkyl amines, glucosyl isothiocyanate, thiocarbamides.

©2014 RASAYAN. All rights reserved

INTRODUCTION

The chemistry of thiourea of carbohydrate is extensively elaborated and well documented¹, many of these derivatives have been found to posses wide applications in industry as carbohydrate based detergent² and in medicine as a anti cancer agents³ and anti fungal agents⁴.

These compounds arouse interest as potential biologically active substances and versatile intermediate for preparing various derivatives. The biological activity of some polyhydroxy compounds have been reported⁵⁻⁶. Glucopyranosyl isocyanate and isothiocyanate⁷, likewise other glycosyl isocyanates have a significant role in synthetic carbohydrate chemistry. Many compounds have been synthesized by the use of glycosyl isothiocyanate⁸⁻¹¹. On the basis of knowledge gained on the work done on *N*-glucosylated compounds, it was quite interesting to synthesize some new *N*-glucopyranosylated thioamides.

Herein, we report the synthesis of 1-tetra-*O*-benzoyl- β -D-glucopyranosyl-3-substituted thiocarbamides by the interaction of tetra-*O*-benzoyl- β -D-glucopyranosyl isothiocyanate and alkyl amines. All the products obtained have been crystallized from ethanol-water. IR spectra of the products show characteristic absorption of prominent peaks, ¹H NMR spectra shows characteristics peaks due to N-H, benzoyl and glucopyranosyl protons while the Mass spectra shows peak¹²⁻¹⁴.

EXPERIMENTAL

General methods

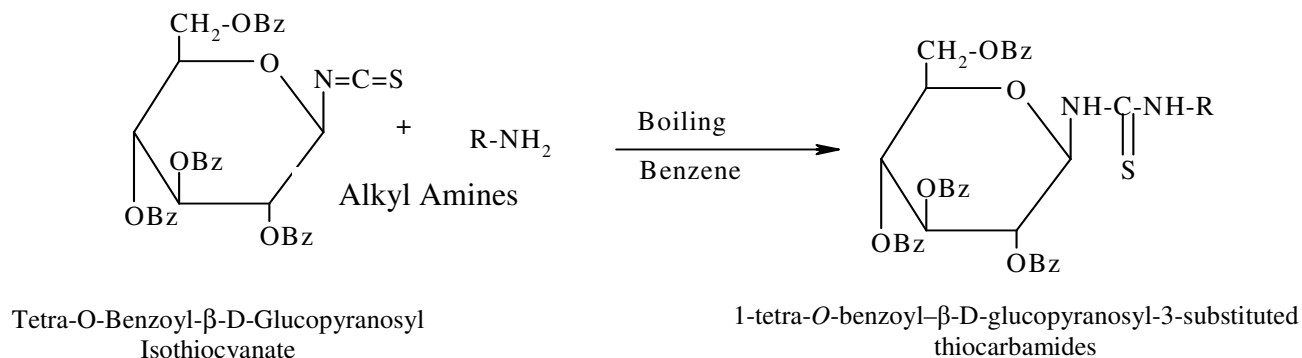
Melting points are uncorrected. Optical rotations [α]_D were measured on an Equip-Tronics digital polarimeter model no. EQ 800 in CHCl₃ at 30°C. IR spectra were recorded on a Perkin-Elmer spectrum RXI (4000-450cm⁻¹) FTIR spectrometer. ¹HNMR were obtained on a Bruker DRX-300 (300 MHz FT NMR) NMR spectrometer for a sample in CDCl₃ solution with TMS as an internal reference. The mass spectra were recorded on a Micromass Quattro II triple quadruple mass spectrometer.

Tetra-*O*-benzoyl- β -D-glucopyranosyl isothiocyanate (2)

Tetra-*O*-benzoyl- β -glucopyranosyl isothiocyanate (2) was prepared by interaction of Tetra-*O*-benzoyl- α -D-glucopyranosyl bromide and lead thiocyanate in anhydrous xylene medium¹⁵.

1-tetra-*O*-benzoyl-β-D-glucopyranosyl-3-methyl thiocarbamides

Condensation of methyl amines (0.005M, 0.95 mL) and tetra-*O*-benzoyl-β-glucopyranosyl isothiocyanate (0.005M, 2.0 g), in benzene (20mL) was carried out on boiling water bath for 4h. The reaction was monitored by TLC, after completion of the reaction, the solvent was distilled off and the resultant sticky residue was triturated with petroleum ether (60-80⁰ C) to afford the solid with m.p. (100-101⁰C). It was crystallized from water-alcohol.



Where, R = (a) methyl, (b) ethyl, (c) n-propyl, (d) ter-butyl, (e) n-pentyl ; Bz = - COC₆H₅.

Scheme-1

RESULTS AND DISCUSSION

Several 1-tetra-*O*-benzoyl-β-D-glucopyranosyl-3-substituted thiocarbamides (3a-e) (Scheme-1) were prepared by the condensation of alkyl amines (1a-e) and tetra-*O*-benzoyl-β-glucopyranosyl isothiocyanate (2) in benzene for 4hrs. The reaction was monitored by TLC. After complete reaction, the solvent was distilled off and the resultant sticky residue was triturated with petroleum ether (60-80⁰ C) to afford the products (3a-e). All the products were crystallized from ethanol-water. Structures of all synthesized products were established on the basis of usual chemical transformations and IR, ¹HNMR and Mass spectral studies.

Table -1: Physical data of 1-tetra-*O*-benzoyl-β-D-glucopyranosyl-3-substituted thiocarbamides (3a-e) (Scheme-1).

S. No.	Reactant (g)	Product	m. p. (°C)	Yield, (%)	[α] _D ³⁹ (c, CHCl ₃)	R _f (3:2, CCl ₄ : EtOAc)	Elemental analysis found (Required)	
							N	S
1	1a (1.21)	3a	90-95	72	122.2 ⁰ (0.9)	0.93	4.09 (4.19)	4.69 (4.79)
2	1b (1.40)	3b	100-105	74.94	-32.5 ⁰ (0.93)	0.85	4.01 (4.10)	4.58 (4.69)
3	1c (1.40)	3c	121-125	82.21	-84.2 ⁰ (0.95)	0.80	3.98 (4.02)	4.42 (4.59)
4	1d (1.40)	3d	115	80.50	85.1 ⁰ (0.94)	0.79	3.82 (3.94)	4.39 (4.50)
5	1e (1.28)	3e	110	75.96	52.5 ⁰ (0.96)	0.90	3.71 (3.86)	4.35 (4.41)

Table -2: Antimicrobial activities of synthesized compounds

Compounds	<i>E. coli</i>	<i>P. vulgaris</i>	<i>S. aureus</i>	<i>P. aeruginosa</i>	<i>A. niger</i>	<i>Penicillium</i>
3a	20	11	18	22	10	12
3b	21	13	15	20	09	14
3c	12	09	15	17	14	16
3d	15	12	13	14	11	17
3e	23	00	17	20	08	09
Amikacin	17	17	17	17	--	--
Fluconazole	--	--	--	--	35	35

Bore size = 6 mm

Antimicrobial Activities

All the compounds have been screened for both anti-bacterial and anti-fungal activities using cup plate agar diffusion method^{16, 17} by measuring the inhibition zone in mm. The compounds were taken at concentration of 1µg/ml using dimethyl sulphoxide as a solvent. Amikacin was used as standard for anti-bacterial activities and Fluconazole as a standard for anti fungal activities. The compounds were screened against various pathogenic bacteria such as *Escherichia coli*, *Proteus vulgaris*, *Staphylococcus aureus*, *Pseudomonas aeruginosa*, *Aspergillus niger* and *Penicillium*. Among synthesized compounds **3a**, **3b** and **3e** show strong activity against *Escherichia coli*, *Pseudomonas aeruginosa* and remaining compounds show good and moderate activity against *Proteus vulgaris*, *Aspergillus niger*, *Staphylococcus aureus* and *Penicillium*. Results are tabulated in Table -2.

1- tetra-*O*-benzoyl-β-D-glucopyranosyl-3-methyl thiocarbamide (3a)

IR (KBr): ν 3360 (N-H); 3061 (Ar-H); 2939 (C-H ali); 1176 (C = S) 1315 (C-N); 846(Char. D-Glucose). **¹HNMR (CDCl₃):** δ 8.23-7.43 (m, 20H Ar-H); 6.2 (d, 1H, NH (glu.)); 5.72-5.03 (m, 7H, glucopyranosyl ring); 1.6-1.3 (s, 3H,CH₃). **ESI Mass (M/z) :** 677.4 (M⁺.+1), 579, 353, 135, 121. Anal. Calcd. for: C₃₅H₃₈O₉ N₃S: C, 62.13; H, 5.62; N, 6.21; S, 4.73, found, C, 56.70; H, 5.40; N, 6.75; S, 5.25, %.

1-tetra-*O*-benzoyl-β-D-glucopyranosyl-3-ethyl thiocarbamide (3b)

IR (KBr): ν 3361 (N-H); 3062 (Ar-H); 2972 (C-H ali); 1176 (C = S) 1315 (C-N); 846(Char. D-Glucose). **¹HNMR (CDCl₃):** δ 8.23-7.43 (m, 20H Ar-H); 6.2 (d, 1H, NH (glu.)); 5.72-5.03 (m, 7H, glucopyranosyl ring); 1.6-1.3 (s, 3H,CH₃). **ESI Mass (M/z) :** 690.4 (Not located), 579, 353, 135, 121. Anal. Calcd. for: C₃₆H₄₀O₉ N₃S: C, 62.03; H, 5.50; N, 6.35; S, 4.70, found, C, 59.50; H, 5.23; N, 6.75; S, 5.65, %.

1- tetra-*O*-benzoyl-β-D-glucopyranosyl-3-n-propyl thiocarbamide (3c)

IR (KBr): ν 3366 (N-H); 3056 (Ar-H); 2942 (C-H ali); 1176 (C = S) 1315 (C-N); 846(Char. D-Glucose). **¹HNMR (CDCl₃):** δ 8.23-7.43 (m, 20H Ar-H); 6.2 (d, 1H, NH (glu.)); 5.72-5.03 (m, 7H, glucopyranosyl ring); 1.6-1.3 (s, 3H,CH₃). **ESI Mass (M/z) :** 706.4 (M⁺.+1), 579, 353, 135, 121. Anal. Calcd. for: C₃₇H₄₆O₉ N₃S: C, 62.13; H, 5.62; N, 6.21; S, 4.73, found, C, 56.70; H, 5.40; N, 6.75; S, 5.25, %.

ACKNOWLEDGEMENTS

Authors are thankful to SAIF, CDRI Lucknow for providing the spectral data. Authors also thanks to Dr. S. G. Bhadange, The Principal Shri Shivaji College, Akola and Dr. S. P. Deshmukh, Head Department of chemistry for encouragement and providing necessary facilities

REFERENCES

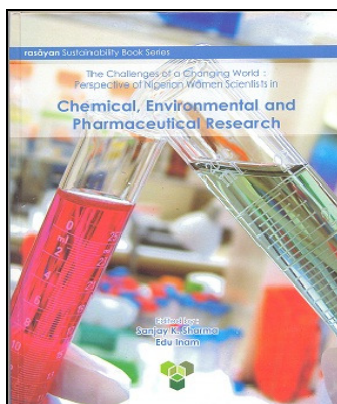
- (a.)I. Goodman, *Adv. Carbohydr. Chem.*, **13** 215(1958); (b) J. M. Gradia-Fernandez, C.O. Mellet, *Adv.Carbohydr. Chem. Biochem.*, **55**, 35(1994); (c) F.O. Caballero, Gimenez Martinez, L. Gracia-Fuentes, E.O.Salmeron, F.Santoyo-Glonzalez, A.V.Berenguel, *J. Org. Chem.*, **66** 7786 (2001).
- J. M. Garcia Fernandez, C. O. Mellet, *Adv. Carbohydr. Chem. Biochem.*, **55** 35(2000).

3. C. Prata, N. Mora, J. M. Lcombe, J. Maurizis, B. Pucci, *Carbohydr. Res.*, **321**, 4(1999).
4. M. M. De Oliveira, M. C. F. Linardi, M. R. P. Sampaio, *J. Pharm. Sci.*, **67**, 562(2006).
5. G. R. Revenkar, P. K. Gupta, A. D. Adams, N. K. Dalley. P.A. McKiernan, P. D. Cook, P. G. Cononico, R. K. Robins, *J. Med. Chem.*, **27**, 1389(1984).
6. Z. J. Witczak, *Adv. Carbohydr. Chem. Biochem.*, **44**, 91(1986).
7. Z. J. Witczak, *Tetrahedron Lett.*, **27**, 155(1986).
8. J. Isac-Garcia, F. G. Calvo-Flores, F. Hernandez-Mateo, F. Santoyo- Gonzalez, *Eur. J. Org. Chem.*, **2**, 383(2001).
9. Y. Gama, I. Shibuya, M. Shimizu, *Chem. Pharm. Bull.*, **50**, 1517(2002).
10. S. P. Deshmukh; B. N. Berad; M. G. Paranjpe, *J. Ind. Chem. Soc.*, **LXIII**, 315(1986). G. V. Korpe; S. P. Deshmukh, *J. Ind. Chem. Soc.*, **79**, 972(2002).
11. (a) R.M. Silverstein; G. C. Bassler; T. C. Morill "Spectrometric identification of Organic Compounds", 5th Ed., John Wiley and Sons, INC, New York, (1991); (b) D.H. Williams and I. Fleming "Spectrometric methods of Organic Chemistry," 4th Ed., Tata-McGraw-Hill publication, (1991); (c) J. R. Dyer, "Applications of absorption spectroscopy of Organic Compounds," 8th Ed., Prentice Hall, (1991).
12. F. Sansone, E. Chierici, A. Casanti, R. Ungaro, *Org. Biomol. Chem.*, **1**, 1802(2003).
13. M. A. Saleh, *Sulfur Lett.*, **25**, 235(2002).
14. K. M. Harring and T. B. Johnson, *J. Am. Chem. Soc.*, **55**, 395 (1933).
15. F. Kawangh, *Analytical Microbiology*, Academic press, New York, (1963).
16. British Pharmacopoeia- II, Biological assay and Tests, The Stationary Office Ltd., London, A-205, (1998).

[RJC-1116/2014]

Absolutely FREE*

[For New life Members]



The Challenges of a Changing World: Perspective of Nigerian Women Scientists in Chemical, Environmental and Pharmaceutical Research

[ISBN: 978-81-921149-0-3]

Print Price: Rs. 990/- only

Be a Proud Life Member of RASĀYAN J. Chem.

Life Membership for Individuals: Rs.8000/- for Indians and USD 1000 for others.**Life Membership for Institutional:** Rs.10000/- for Indians and USD 1500 for others.**BENEFITS OF LIFEMEMBERSHIP:**

1. You will receive the journal and all its special issues regularly lifelong.
2. If you are a LIFE MEMBER, you need not to pay subscription fee every time for publication of your paper in RJC.
3. You'll be a Reviewer for RJC manuscripts of your Field Interest and we'll publish your name in our journal.
4. You will be exempted from Registration Fee of any National or International future events (i.e. workshop, seminars, Conferences etc.) organized by RJC.
5. You may be elected as Editorial Member of RJC (Note: It'll depend upon your publication and scientific achievements).
6. New Life members shall have a **BOOK*** absolutely **FREE** from RJC with Complements.

For being a **Life Membership**, just mail to editor-in-Chief with your detailed Resume.

Correspondence address:

23 'Anukampa', Janakpuri, Opp. Heerapura Power Stn., Ajmer Road, Jaipur-302024 (India)
E-mail : rasayanjournal@gmail.com ; Phone : 0141-2810628(Off.), 07597925412(Mob.)